

**Crystal Data:** Orthorhombic. *Point Group:*  $2/m\ 2/m\ 2/m$ . Short prismatic crystals, to 2 mm, with large {101}, {111}, modified by {301}, {100}, {110}, {230}, {010}.

**Physical Properties:** *Cleavage:* Perfect on {101}. *Hardness* = 3 *D*(meas.) = 4.77(5)  
*D*(calc.) = 4.887

**Optical Properties:** Transparent. *Color:* Bluish green, deep green; bluish green in transmitted light. *Luster:* Vitreous.

*Optical Class:* Biaxial (-). *Pleochroism:* *X* = colorless; *Y* = pale bluish green; *Z* = bluish green. *Orientation:* *X* = *c*; *Y* = *b*; *Z* = *a*. *Dispersion:*  $r > v$ , extreme.  $\alpha = 1.786(5)$   $\beta = 2.070(10)$   
 $\gamma = 2.075(10)$   $2V(\text{meas.}) = 0^\circ\text{--}5^\circ$

**Cell Data:** *Space Group:*  $Pnma$ .  $a = 10.794(2)$   $b = 6.708(1)$   $c = 4.781(1)$   $Z = 4$

**X-ray Powder Pattern:** Chuquicamata, Chile. (ICDD 22-236).  
 4.37 (100), 3.66 (65), 2.394 (60), 1.788 (55), 2.698 (50), 5.38 (40), 2.636 (30)

Chemistry:	(1)	(2)
I <sub>2</sub> O <sub>5</sub>	64.79	65.33
CuO	30.62	31.14
Na <sub>2</sub> O	0.59	
H <sub>2</sub> O	3.68	3.53
Total	99.68	100.00

(1) Chuquicamata, Chile; H<sub>2</sub>O + I<sub>2</sub> by the Penfield method, then I<sub>2</sub> dissolved by KI and titrated, H<sub>2</sub>O taken by difference; corresponds to (Cu<sub>0.99</sub>Na<sub>0.04</sub>)<sub>Σ=1.03</sub>(IO<sub>3</sub>)<sub>1.00</sub>(OH)<sub>1.05</sub>. (2) Cu(IO<sub>3</sub>)(OH).

**Occurrence:** A rare mineral in the oxidized zone of a copper porphyry deposit.

**Association:** Kaolin, quartz.

**Distribution:** From Chuquicamata, Antofagasta, Chile.

**Name:** To honor Reno H. Sales (1876–1969), Chief Geologist of the Anaconda Company, responsible for initial development of the mine at Chuquicamata, Chile.

**Type Material:** Harvard University, Cambridge, Massachusetts, USA, 95027.

**References:** (1) Palache, C., H. Berman, and C. Frondel (1951) Dana's system of mineralogy, (7th edition), v. II, 315–316. (2) Ghose, S. and C. Wan (1978) Salesite, CuIO<sub>3</sub>(OH), and Cu(IO<sub>3</sub>)<sub>2</sub>•2H<sub>2</sub>O: a comparison of the crystal structures and their magnetic behavior. *Amer. Mineral.*, 63, 172–179.