

**Crystal Data:** Monoclinic. *Point Group:*  $2/m$ . As prismatic crystals with nearly square cross sections, to 50 cm; granular, columnar, lamellar massive. *Twinning:* Simple or multiple twins on {100} or {010}, common.

**Physical Properties:** *Cleavage:* Distinct on {110}, (110)  $\wedge$  (1 $\bar{1}$ 0)  $\sim$ 87°; partings on {100} and probably {010}. *Fracture:* Uneven to conchoidal. *Tenacity:* Brittle. Hardness = 5.5–6.5 D(meas.) = 3.22–3.38 D(calc.) = 3.278

**Optical Properties:** Transparent to opaque. *Color:* Colorless, white, yellow, pale to dark green, black; colorless in thin section. *Streak:* White, gray, gray-green. *Luster:* Vitreous or dull.

*Optical Class:* Biaxial (+). *Orientation:*  $Y = b$ ;  $Z \wedge c = -38^\circ$  on (010);  $X \wedge a = -22^\circ$ .  
*Dispersion:*  $r > v$ , weak to moderate.  $\alpha = 1.664$   $\beta = 1.672$   $\gamma = 1.694$   $2V(\text{meas.}) = 59^\circ$

**Cell Data:** *Space Group:*  $C2/c$ .  $a = 9.746$   $b = 8.899$   $c = 5.251$   $\beta = 105.63^\circ$   $Z = 4$

**X-ray Powder Pattern:** Schwartzstein, Austria. (ICDD 11-654).  
2.991 (100), 2.528 (40), 2.893 (30), 2.518 (30), 3.23 (25), 2.952 (25), 1.625 (25)

<b>Chemistry:</b>	(1)	(2)	(1)	(2)	(1)	(2)	
SiO <sub>2</sub>	54.66	54.09	FeO	0.07	1.47	K <sub>2</sub> O	0.15
TiO <sub>2</sub>		0.28	MnO	0.02	0.09	H <sub>2</sub> O <sup>+</sup>	0.22
Al <sub>2</sub> O <sub>3</sub>	0.07	1.57	MgO	18.78	16.96	H <sub>2</sub> O <sup>-</sup>	0.08
Fe <sub>2</sub> O <sub>3</sub>	0.68	0.74	CaO	25.85	21.10	rem.	0.49
Cr <sub>2</sub> O <sub>3</sub>		2.03	Na <sub>2</sub> O		1.37		
						<b>Total</b>	<b>100.35</b> <b>100.64</b>

(1) Juva, Finland; corresponds to Ca<sub>1.00</sub>(Mg<sub>1.01</sub>Fe<sub>0.02</sub><sup>3+</sup>)<sub>Σ=1.03</sub>Si<sub>1.98</sub>O<sub>6</sub>. (2) Dutoitspan mine, Kimberley, Cape Province, South Africa; corresponds to (Ca<sub>0.82</sub>Na<sub>0.05</sub>Fe<sub>0.04</sub><sup>2+</sup>Mg<sub>0.04</sub>K<sub>0.01</sub>)<sub>Σ=0.96</sub>(Mg<sub>0.88</sub>Cr<sub>0.06</sub>Al<sub>0.03</sub>Fe<sub>0.02</sub><sup>3+</sup>Ti<sub>0.01</sub>)<sub>Σ=1.00</sub>(Si<sub>1.96</sub>Al<sub>0.04</sub>)<sub>Σ=2.00</sub>O<sub>6</sub>.

**Polymorphism & Series:** Forms two series, with hedenbergite, and with johannsenite.

**Mineral Group:** Pyroxene group.

**Occurrence:** Typical of metamorphosed siliceous Ca, Mg-rich rocks of the pyroxene-hornfels or epidote-amphibolite facies; common in skarns, Ca, Mg-rich gneisses and schists, and some kimberlites and peridotites. Less common in alkalic olivine basalts and andesites.

**Association:** Calcite, forsterite, chondrodite, monticellite, clinohumite, scapolite, wollastonite, grossular, vesuvianite, tremolite, quartz.

**Distribution:** Selected localities for fine crystals follow: at Schwarzenstein, Zillertal, and near Prägraten, Tirol, Austria. From Ala, Piedmont, and St. Marcel, Val d'Aosta, Italy. At Otokumpu, Finland. In Russia, at the Akhmatovsk deposit, near Zlatoust, Ural Mountains; large crystals in the Inagli massif, 30 km west of Aldan, Yakutia; and along the Slyudyanka River, near Lake Baikal, Siberia. In Canada, many localities; in Ontario, at Bird's Creek, Eganville, Dog's Lake, Littlefield, and Burgess; in Quebec, at Wakefield, Brompton Lake, near Magog, and in the Jeffrey mine, Asbestos. In the USA, at DeKalb, St. Lawrence Co., Natural Bridge, Jefferson Co., Sing Sing, near Ossining, Westchester Co., New York; and at Ducktown, Polk Co., Tennessee. At Ampandrandava and Andranodambo, Taolainaro (Fort Dauphin), Madagascar. Large gemmy crystals from the Kunlun Mountains, Sinkiang Uighur Autonomous Region, China. From Tange-Achin, Kandahar Province, Afghanistan. Found near Jaipur, Rajasthan, India. At Khapalu and Chamachu, Pakistan.

**Name:** From the Greek for *double* and *appearance*, apparently for two possible orientations of the prism zone.

**References:** (1) Dana, E.S. (1892) Dana's system of mineralogy, (6th edition), 351–359. (2) Deer, W.A., R.A. Howie, and J. Zussman (1978) Rock-forming minerals, (2nd edition), v. 2A, single-chain silicates, 198–293.

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